

LTS Research Laboratories, Inc. Safety Data Sheet Copper Sulfate (anhydrous)

1. Product and Company Identification	
Trade Name:	Copper sulfate
Chemical Formula:	$CuSO_4$
Recommended Use:	Scientific research and development
Manufacturer/Supplier:	LTS Research Laboratories, Inc.
Street:	37 Ramland Road
City:	Orangeburg
State:	New York
Zip Code:	10962
Country:	USA
Tel #:	855-587-2436 / 855-lts-chem
24-Hour Emergency Contact:	800-424-9300 (US & Canada)
	+1-703-527-3887 (International)
	2. Hazards Identification
Signal Word:	Warning
Hazard Statements:	H302 Harmful if swallowed
	H315 Causes skin irritation H319 Causes serious eye irritation
	H400 Very toxic to aquatic life
Precautionary Statements:	P280 Wear protective clothing/equipment
	P273 Avoid release to the environment
	P305+P351 If in eyes: Rinse cautiously with water for several minutes P301+P312 If swallowed: Call Poison Center or doctor.
HMIS Health Ratings (0-4):	
Health:	2
Flammability:	0
Physical:	0
	3. Composition
Chemical Family:	Salt
Additional Names:	Copper sulfate anhydrous, Cupric sulfate, Copper(II) sulfate
Copper sulfate (CuSO ₄):	
Percentage:	100 wt%
CAS #:	7758-98-7
EC #:	231-847-6

4. First Aid Procedures
Seek medical attention if symptoms persist.
None
None
Remove victim to fresh air. Supply oxygen if breathing is difficult.
Give one to two glasses of water and induce vomiting. Never induce
vomiting or give anything by mouth to an unconscious person. Wash affected area with mild soap and water. Remove any
contaminated clothing.
Flush eyes with water, blinking often for several minutes. Remove contact lenses if present and easy to do. Continue rinsing
5. Firefighting Measures
Non-flammable
No special restrictions – use suitable extinguishing agent for surrounding material and type of fire.
Use full-face, self-contained breathing apparatus with full protective
clothing to prevent contact with skin and eyes. See section 10 for
decomposition products.
6. Accidental Release Measures
Wear appropriate respiratory and protective equipment specified in
special protection information. Isolate spill area and provide
ventilation. Vacuum up spill using a high efficiency particulate
absolute (HEPA) air filter and place in a closed container for disposal Take care not to raise dust.
Isolate runoff to prevent environmental pollution.
7. Handling and Storage
Wash thoroughly after handling.
Store in a cool dry place in a tightly sealed container. Store apart from
materials and conditions listed in section 10.
Do not use tobacco or food in work area. Wash thoroughly before
eating and smoking. Do not blow dust off clothing or skin with
compressed air.
Provide sufficient ventilation to maintain concentration at or below threshold limit.
Exposure Controls and Personal Protection
None
None
None
Dust Respirator
Rubber gloves
Safety glasses or goggles
Protective work clothing. Wear close-toed shoes and long

9. P	hysical and Chemical Characteristics
Color	Pale blue
Form:	Powder, Granules, Pellets, Sputtering target, Custom parts
Odor:	Odorless
Water Solubility:	forms hydrate, dissolves 203 g/l
Boiling Point:	N/A 200 °C
Melting Point: Flash Point:	N/A
Autoignition Temperature:	N/A
Density:	3.603 g/cc
Molecular weight:	159.61 g/mol
	10. Reactivity
Stability:	Stable under recommended storage conditions
Reacts With:	Oxidizing agents
Incompatible Conditions:	None
Hazardous Decomposition Products:	Metal oxide fume
	11. Toxicological Information
Potential Health Effects:	
Eyes:	Eye contact with copper sulfate can cause conjunctivitis, inflammation
C1 :	of the eyelid lining, ulceration, and clouding of the cornea.
Skin: Ingestion:	Itching or eczema Moderately toxic. According to studies, the lowest dose of copper
	sulfate that had a toxic impact on humans is 11 mg/kg. Because of its irritating effect on the gastrointestinal tract, vomiting is automatically triggered in case of the ingestion of copper sulfate. However, if copper sulfate is retained in the stomach, the symptoms can be severe. After 1-12 grams of copper sulfate are swallowed, such poisoning signs may occur as a metallic taste in the mouth, burning pain in the chest, nausea, diarrhea, vomiting, headache, discontinued urination, which leads to yellowing of the skin. In cases of copper sulfate poisoning, injury to the brain, stomach, liver, or kidneys may also occur.
Chronic:	Copper compounds may be irritating to the skin, eyes and respiratory tract. They may cause metal fume fever, hemolysis of the red blood cells and injury to the liver, lungs, kidneys and pancreas. Ingestion may also cause vomiting, gastric pain, dizziness, anemia, cramps, convulsions, shock, coma and death
Signs & Symptoms:	N/A
Aggravated Medical Conditions:	N/A N/A
Median Lethal Dose:	300 mg/kg for rat by mouth
Carcinogen:	N/A
	12. Ecological Information
Aquatic Toxicity:	High
Persistent Bioaccumulation Toxicity:	No
Very Persistent, Very Bioaccumulative:	No
Notes:	Danger to drinking water even in small quantities. May have long- lasting harmful effects to aquatic life.

	al, and international regulations. 4. Transportation Data 2. ardous for transportation
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Hazardous: Haz	zardous for transportation
Hazard Class: 9 M	liscellaneous hazardous substances
Packing Group: III	
UN Number: UN	3077
Proper Shipping Name: Env	rironmentally hazardous substance, solid, n.o.s. (Copper sulfate)
15.	Regulatory Information
Sec 302 Extremely Hazardous: No	
Sec 304 Reportable Quantities: N/A	Λ
Sec 313 Toxic Chemicals: Yes	3
1	6. Other Information

given is based on our knowledge of this product, at the time of publication. It is given in good faith. The attention of the user is drawn to the possible risks incurred by using the product for any other purpose other than that for which it was intended. This does not in any way excuse the user from knowing and applying all the regulations governing his activity. It is the sole responsibility of the user to take all precautions required in handling the product. The aim of the mandatory regulations mentioned is to help the user to fulfill his obligations regarding the use of hazardous products.

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